# DESIGN DISTILLATION COLUMN FOR SEPARATION OF ORTHO, META, PARA NITRO TOLUENE AND OPTIMIZATION USING ASPEN HYSYS 3.2

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#### ABSTRACT

Aspen Hysys 3.2 is simulation software used for simulation of chemical process .It is market leading process modeling tool for conceptual design, optimization, asset management & performance monitoring .Distillation is one the most common industrial separation process .Efforts have been made by researchers to various design method, minimize energy consumption, optimization and capital cost by inventing analytical and graphical design method. This paper shows the case study for design multi component distillation column Ortho Nitro Toluene, Meta Nitro Toluene & Para Nitro Toluene by analytical method as well Aspen hysys 3.2.

Keywords : Aspen HYSYS, Multi component Distillation, Isomer Separation, Packed column, simulation of distillation column. I.INTRODUCTION

Table	1:	<b>Preliminary</b>	calculation	Data
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	ONT	MNT	PNT
Molecular Weight	137.13	137.13	137.13
Boiling point °C	221.7	232.6	238.3
Specific Gravity	1.162	1.286	1.157

#### **Design of the Distillation column**

We have mixture of Ortho, Meta, Para Nitro Toluene which has to be separated at purity of MNT more than 95 %, ONT & PNT should be more than 99 %. Flow rate of the feed is 25000 kg/hr which contains mole fraction of ONT = 60% (15000 Kg/hr), MNT = 6% (1500Kg/hr), PNT = 34% (8500 Kg/hr)



Table 2 Calculation of vapor pressure using Antoine constant at 140  $^{0}$  C

	ONT	MNT	PNT
T <sup>0</sup> C	140.3	140.3	140.3
$\frac{B}{T+C}$	5.853426	6.112369	6.437824
$\frac{A-B}{T+C}$	1.837944	1.752971	1.664856
mmHg(Abs)	68.86	56.62	46.22

Table 3 Relative volatility of component of feed to column No-T 101

Relative Volatility (O/M)	$\frac{68.85}{56.62} = 1.21$
Relative Volatility (M/P)	$\frac{56.62}{46.22}$ = 1.22
Relative Volatility (O/P)	$\frac{68.85}{46.22} = 1.48$

Geometric Average of relative volatility (used for FENKES's equation)

Bubble point was calculated as follows

$$\mathbf{Y}_{a} = \frac{\mathbf{X} \mathbf{a} \, \mathbf{P} \mathbf{a}^{\circ}}{\mathbf{T} \mathbf{P}}$$

Where,

 $Y_a$ = Vapor pressure ,  $X_a$ = Mole Fraction ,  $Pa^\circ$  =Partial Pressure of individual component

 $T_p = Total pressure$ 

Table 4: Bubble point calculation

		Feed mole fraction	Re boiler pressure (abs mmHg)
ONT	0.68856	0.6	60
MNT	0.05662	0.06	60
PNT	0.261929	0.34	60

Corresponding bubble point with 60 mmHg pressure at 140 °C Mole fraction was calculated.

Table 5 :	Mole	fraction of	f the	distillate	and bottom	of	column	No-	T 1	<i>01</i>
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	ONT	MNT	PNT	Total
Distillate mole	0.995	0.005	0	1
fractions				
Bottom mole	0.005	0.1429	0.8621	1
fraction				

Now we select light key and heavy component

Table 6 : Heavy	y and light keys componer	nt in distillate and bottom	of T 101
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Light key in distillate	O N T
Heavy key in distillate	M N T
Heavy key in Bottom	P N T
Light key in Bottom	M N T

Calculation of minimum no of stages

Ratio of light key to heavy key component in distillate  $\frac{0.995}{0.005} = 199$ 

Ratio of heavy to light key component in bottom

$$\frac{0.8521}{0.1429}$$
=5.96

Minimum no of stages can be calculated as

 $=(\frac{\log (5.96 \times 199)}{\log mean (Relative volatility of at bubble temperature )}) =\frac{7.07}{0.3}$ 

= Minimum No of stages are 26.94 = 27

FUG Method does not give information about feed tray location .feed tray location can be determine by using following equation (Kirkbride Equation)

$$\log \left\{ \frac{Nr}{Ns} \right\} = \left[ 0.206 \log \left( \frac{B}{D} \right) \left( \frac{X,HK}{X,LK} \right)_{f} \left( \frac{Xb,LK}{Xd,HK} \right)^{2} \right]$$
$$N_{r} = 12 \qquad N_{s} = 15$$

Using *Underwood*'s second equation (at  $q \approx 1$ ):

$$1 - q = \sum_{i=1}^{\frac{\alpha_i x_{if}}{\alpha_i - \theta}} \tag{1}$$

Now here onwards we need to solve Underwood equation which will give value of 0.88 or nearby,

For that we need to solve Underwood equation assuming  $\theta = 2.82$ 

E.g. for O N T =
$$(\frac{1.22 \times 0.6}{1.2 - 2.8})$$
 = -0.4549

### Table 7 Value of $X_f$ for Underwood equation

ONT	MNT	PNT	Total
0.45	-0.45	-0.37	-0.04

$$\left(\frac{L}{D}\right)_{\min} = \left(\frac{1 - \left\{(\sum 0, P, M \text{ (Mole fraction in distillate} \times \alpha)\right\}}{(\alpha - \theta)}\right)$$

After getting value of  $\theta$  now we have to calculate value for  $(\frac{L}{\rho})_{\min}$ 

# Table 8: Calculation of $(\frac{L}{D})_{min}$ for each component.



### =1-(-0.75) =1.75

Now calculating optimum reflux ratio

=1.2 ×1.75

Optimum Reflux ration=2.10

## Design of 2<sup>nd</sup> column T 102

# Table 9 Calculation of vapor pressure using Antoine constant at 150 C $^{0}$

	ONT	MNT	PNT
$\mathbf{T}  (\mathbf{C}^{0})$	150.353	150.353	150.353
$\frac{B}{T+C}$	5.702224	5.964183	6.2879

$\frac{A-B}{T+C}$	1.989146	1.901157	1.8147
mmHg	97.53182	79.64466	65.2753

Table 10 Relative volatility of component of feed to column No-T 102

Relative Volatility (O/M)	1.224587006
Relative Volatility (M/P)	1.220134666
Relative Volatility (O/P)	1.494161057

Geometric Average of relative volatility (used for FENKES's equation )

=1.30

Bubble point was calculated as follows

$$Y_a = \frac{Xa Pa^{\circ}}{TP}$$

Where,

 $Y_a$ = Vapor pressure ,  $X_a$ = Mole Fraction ,  $Pa^\circ$  =Partial Pressure of individual component

 $T_p = Total pressure$ 

### Table 11 : Mole fraction of feed to column No- T 102

		Feed mole fraction	Re boiler pressure (abs mmHg)
ONT	0.007225	0.005	67.5
MNT	0.168611	0.1429	67.5
PNT	0.824016	0.8521	67.5

Table 12 : Mole fraction of the distillate and bottom of column No- T 102

		ONT	MNT	PNT	Total
Distillate fractions	mole	0.033	0.940	0.027	1
Bottom fraction	mole	0.000	0.00056	0.99943	1

Now we select light key and heavy component

#### Table 13: Heavy and light keys component in distillate and bottom of T 102

Light key in distillate	ΜΝΤ
Heavy key in distillate	P N T
Heavy key in Bottom	P N T
Light key in Bottom	M N T

#### Calculation of minimum no of stages

Ratio of light key to heavy key component in distillate  $\frac{0.94}{0.027} = 34.81$ 

Ratio of heavy to light key component in bottom  $, \frac{0.9994}{0.00056} = 1784.64$ 

Minimum no of stages can be calculated as

=( $\frac{\log \mathbb{E}34 \times 1784.64)}{\log mean (Relative volatility of at bubble temperature )}$ )

= Minimum No of stages are 41.24 = 42

FUG Method does not give information about feed tray location .feed tray location can be determine by using following equation (Kirkbride Equation)

$$Log \left\{ \frac{Nr}{Ns} \right\} = \left[ 0.206 \log \left( \frac{B}{D} \right) \left( \frac{X,HK}{X,LK} \right)_{f} \left( \frac{Xb,LK}{Xd,HK} \right)^{2} \right]$$
$$N_{r} = 7 \qquad N_{s} = 35$$

Using *Underwood*'s second equation (at  $q \approx 1$ ):

$$1 - q = \sum_{i=1}^{\frac{\alpha_i x_{if}}{\alpha_i - \theta}} \tag{1}$$

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Now here onwards we need to solve Underwood equation which will give value of 0.88 or nearby,

For that we need to solve Underwood equation assuming  $\theta = 2.51$ 

E.g. for O N T = $(\frac{1.22 \times 0.005}{1.22 - 2.51})$  = -0.0047

### Table 14 Value of $X_f$ for Underwood equation

ONT	MNT	PNT	Total
-0.004754357	-0.13	-1.25	-1.38

 $\left(\frac{L}{D}\right)_{\min} = \left(\frac{1 - \left\{\left(\sum O, P, M \text{ (Mole fraction in distillate } \times \alpha\right)\right\}}{(\alpha - \theta)}\right)$ 

After getting value of  $\theta$  now we have to calculate value for  $(\frac{L}{D})_{\min}$ 

Table 15: Calculation of  $(\frac{L}{p})_{\min}$  for each component.

ONT	MNT	PNT	Total
-0.0323	-0.907	-0.0073	-0.946
$(\frac{L}{2})$ . $-(\frac{1-\{(\sum 0, P, M)\}}{2})$	Mole fraction in distillate× $\alpha$ )	<u>}</u>	
$D^{min} = C$	(α-θ)	)	

=1-(-0.946) =1.946

Now calculating optimum reflux ratio

=1.2 ×1.946

=2.33

R+1=3.33

Optimum reflux ration =3.33

Now same column we will simulate using Aspen Hysys



Figure 2: Column pictorial

Note -NRTL model is used to simulate column

Table 16: Showing column performance through HYSYS

Column Parameter	Result
Minimum no of trays	40
Actual no of trays	80
Optimal feed stages	28.7
Condenser Temperature (°C)	136
Re boiler Temperature (°C)	150

 Table 17: Shows Distillate & Bottom of column No- T 101.
 Particular

	Flow rate	ONT	MNT	PNT
	Kg/Hr	Wt %	Wt %	Wt %
Feed	9974	0.50	14.29	85.21
Distillate	1511.2121	3.3	94.00	2.7
Bottom	8462.7879	0.00	0.056	99.94

Table	18:	Shows	Distillate	& Bottom	of	column	No-	T 102	2.
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Component	Mass Fraction	Flow rate	Mass Fraction	Flow rate
Name	Name in (Kg/H		in	(Kg/Hr)
	Distillate		Bottom	
ONT	0.3	43.17	0	0
MNT	0.01	1381.44	0.005	42.675
PNT	0.96	7.195	0.995	8492.33

### **Optimization of Feed point location through HYSYS of column T 101**

Three different location of feed point were chosen for the feed location optimization

 Table 19: Shows comparison various feed point location of column No- T 101

	Feed stage	10	29	65
Top of T101	Temperature( <sup>0</sup> C)	137.1	136.8	137.7
	Pressure (Kpa)	8.00	8.00	8.00
	Mass Fract. ONT	0.92	0.9634	0.9783
	Mass Fract.MNT	0.0262	0.0197	0.0217
	Mass Fract. PNT	0.0480	0.0169	0.000
	Mass flow (Kg/Hr)	14950	15000	15020

Bottom of T101	Temperature( <sup>0</sup> C)	148.9	149.7	150.
	Pressure(Kpa)	9.00	9.00	9.00
	Mass Fract ONT	0.1150	0.0550	0.0308
	Mass Fract.MNT	0.1103	0.1204	0.1179
	Mass Fract. PNT	0.7747	0.8246	0.8517
	Mass flow (Kg/Hr)	10050	10000	9981
Rectification Stages		10	29	65
Stripping stages		69	50	14

If we see above 3 results it shows that as we goes to increase the stages in rectification section get more purity of ONT in distillate.

## **Optimization of Feed point location through HYSYS of Column T102**

Three different location of feed point were chosen for the feed location optimization

Table 20: Shows comparison various feed point location of column No- T 102

	Feed stage	40	72	95
Top of T102	Temperature( <sup>0</sup> C)	142.1	142.0	142.0
	Pressure (Kpa)	8.00	8.00	8.00

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	Mass Fract. ONT	0.0390	0.0390	0.0387
	Mass Fract.MNT	0.9570	0.9610	0.9238
	Mass Fract. PNT	0.0040	0.0000	0.0375
	Mass flow (Kg/Hr)	1450	1440	1490
Bottom of T102	Temperature( <sup>0</sup> C)	151	151	151
	Pressure(Kpa)	9	9	9
	Mass Fract ONT	0	0	0
	Mass Fract.MNT	0.0114	0.0118	0.0128
	Mass Fract. PNT	0.9969	0.9964	0.9872
	Mass flow (Kg/Hr)	8525	8535	8485
Rectification Stages		40	72	95
Stripping stages		68	36	13

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#### Mass balance across column T 101

Table 2 Showing mass balance at Column No T-101

	ONT	MNT	PNT	Total
Feed to T 101	15000	1500	8500	25000
Distillate of T 101	14699.93	318.100	0	15018
Bottom of T 101	49.87	1424.85	8499.28	9974

(all value are in Kg/hr)

#### Mass balance across column T 102

#### Table 3 Showing mass balance at Column No T-102

	ONT	MNT	PNT	Total
Feed to T 102	49.87	1424.85	8499.28	9974
Distillate of T 102	49.87	1382	0	1431.87
Bottom of T 102	0	42.84	8499.28	8542.13

(all value are in Kg/hr)

Column diameter was calculated for both the column.

## Table 20Diameter for column No T-101 & T- 102

	Rectifying Section	Stripping Section
Column No T-101	2.84	2.79
Column No T-102	1.80	1.76

Tray spacing was 0.5 m

## **II.RESULT**

	Column T101		Column T102	
		By Aspen		By Aspen
	By calculation	Hysys	By Calculation	Hysys
Minimum number of				
stages required	27	40	42	57
Top temperature <sup>o</sup> C	120	136.6	125	142

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Bottom				
temperature <sup>o</sup> C	140	150.3	150.35	151.2
Column top				
pressure(mmHg)abs	32	64	35	60.02
Column bottom				
pressure(mmHg)abs	60	120	67.5	67.52

#### **III.CONCLUSION**

Initially manual calculation was done for the column which took lot of time, but later optimization was done on HYSYS software which gave instant and good results proving to point that designing through HYSYS gives saves lot of time. Through HYSYS we manage to get Ortho and Para fraction at purity of 99.50 % while Meta was obtained at purity of 96.55 % . Both column was operated under vacuum. And decided use same column model with addition of one more batch distillation column for further purification of Top material of column T102.

#### REFERENCES

- 1. K. B. Jana, A. K. and A. N. Samanta, "A novel intensified heat integration in multicomponent distillation", 2012, Energy 41, 443–453.
- Monroy-Loperena, R. & Vacahern, M. A simple, reliable and fast algorithm for the simulation of multicomponent distillation columns. *Chem. Eng. Res. Des.*1, 389–395 (2012).
- Errico, M., Rong, B. G., Tola, G. & Turunen, I. A method for systematic synthesis of multicomponent distillation systems with less than N-1 columns. *Chem. Eng. Process. Process Intensif.*48, 907–920 (2009).

- Yildiz, U., Gurkan, U., Ozgen, C. & Leblebicioglu, K. State Estimator Design for Multicomponent Batch Distillation Columns. *Chem. Eng. Res. Des.*83, 433–444 (2005).
- 5. Pelkonen, S., Górak, a., Ohligschläger, a. & Kaesemann, R. Experimental study on multicomponent distillation in packed columns. *Chem. Eng. Process.***40**, 235–243 (2001).
- Mori, H., Ibuki, R., Taguchi, K., Futamura, K. & Olujić, Ž. Three-component distillation using structured packings: Performance evaluation and model validation. *Chem. Eng. Sci.***61**, 1760–1766 (2006).
- Bessa, L. C. B. a, Batista, F. R. M. & Meirelles, A. J. a. Double-effect integration of multicomponent alcoholic distillation columns. *Energy*45, 603–612 (2012).
- 8. Fatoorehchi, H. & Abolghasemi, H. Journal of the Taiwan Institute of Chemical Engineers Approximating the minimum reflux ratio of multicomponent distillation columns based on the Adomian decomposition method. **45**, 880–886 (2014).
- Baratti, R., Bertucco, A., Da Rold, A. & Morbidelli, M. A composition estimator for multicomponent distillation columns - development and experimental test on ternary mixtures. *Chem. Eng. Sci.*53, 3601–3612 (1998).
- 10. Franklin, N. L. & Forsyth, J. S. The interpretation of minimum reflux conditions in multicomponent distillation. *Chem. Eng. Res. Des.***75**, S56–S81 (1997).
- Cortez-Gonzalez, J. *et al.* Optimal design of distillation systems with less than N-1 columns for a class of four component mixtures. *Chem. Eng. Res. Des.*90, 1425–1447 (2012).
- 12. Process Simulation control Text book By Amiya K Jana IIT khargpur.