

Antimicrobial investigation of silver doped titanium dioxide nanoparticle

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ABSTRACT

Silver doped titanium dioxide nanoparticles were synthesized by chemical reduction method. The prepared sample was characterized by UV-vis spectroscopy, Fourier transform infrared spectroscopy (FTIR), Transmission electron microscopy (TEM), X-ray diffraction (XRD). UV-visible absorption spectra of Ag doped TiO₂ a nanoparticle has been observed at 435nm. FTIR spectra of silver doped titanium dioxide nanoparticles have absorption band at 1550cm⁻¹ corresponding to C=C and C-H group, band at 1678 cm⁻¹, 2262 cm⁻¹, 2402 cm⁻¹, 3702 cm⁻¹ shows stretching of C-H, C=C, C=N, O-H, N-H groups respectively. The antibacterial activity of the sample was studied against bacterial strain: *Pseudomonas aeruginosa* by indicating Zone of inhibition by performing disc diffusion method. *Pseudomonas aeruginosa* is a gram negative bacterium and is multidrug resistant as well as antibiotic resistant that causes severe infection in human body and soft rot disease in plants. The results showed significant antibacterial activity against bacterial strain

Keywords: Doping, TiO₂, Silver, FTIR

1. INTRODUCTION

Silver as a metal nanoparticles have been extensively used as an antipathogens. The antimicrobial property of silver in its ionized state has been explored since time immemorial. But the nanosized of silver enhance its biological properties and make it more effective [1]. Not only as metal nanoparticles, silver along with some metal oxide as well as other materials have been tremendously used in various fields.[2-5]. Silver silica nanocomposites show effective antimicrobial properties against *E.coli* and *S.aureus* [6]. As we are very much aware of polymers and their excellent properties, they can be easily being incorporated in the matrix of polymer to form nanocomposites [7]. Besides others Titanium dioxide nanoparticles is one of the most widely studied semi-conducting photocatalysts for the degradation of organic contaminants from water and air, because of its physical and chemical stability, high catalytic activity, high oxidative power, low cost and ease of production [8–11]. In recent years TiO₂ has been used widely for the preparation of different types of nanomaterials, including nanoparticles, nanorods, nanowires, nanotubes, and mesoporous and nanoporous TiO₂ – containing materials [12]. Doping of TiO₂ nanoparticles with silver is usually done to create heterojunction which ranges

from 10 to 100 nm in size in between the surface of silver in metallic form and that of TiO₂ nanoparticles [13-15].

Titanium dioxide (TiO₂), particularly in its nanoscale range, has evolved a great deal of interest over decades, as it has plentiful potential applications. TiO₂ nanomaterials are of interesting in a broad range of applications such as photocatalysis, dye sensitized solar cells, gas sensors, photochromic devices, photo degradation of organic compounds, deactivation of microorganisms, organic synthesis and cells culture [16]. A number of researches have observed it to be an effective light activated photocatalyst, with strong bactericidal activity. However, its main disadvantage is its wide band-gap, meaning it is only really bactericidal when it absorbs UV light [17].

Silver modification of TiO₂ reduces the band gap energy, allowing visible light to activate the material's photocatalytic activity [18]. Even small quantities of Ag-TiO₂ nanoparticles (>2%wt) are able to produce a significant bactericidal effect when in contact with *S. mutans* under visible light [19].

TiO₂ NPs combined with silver (Ag) in the presence of UV light have been showing greater effect against the growth of *E. coli* compared with TiO₂ NPs without silver [20].

Here in this work we have synthesized Titanium dioxide nanoparticles and doped it with silver to enhance its physical and biological properties. Titanium dioxide nanoparticles itself has significant antimicrobial activity, but silver doping has much enhanced antibacterial activity against many pathogenic bacteria.

2. MATERIAL AND METHOD

2.1 Selection of pathogen

Pseudomonas aeruginosa MTCC number 10235 was purchased from MTCC Chandigarh in freeze dried condition and revived the culture in nutrient agar plates.

2.2 Synthesis of silver doped titanium dioxide nanoparticle

Synthesis of silver doped titanium dioxide nanoparticles can be done by several methods, here we have synthesized by using sodium borohydride as reducing agent at low temperature.

0.2gm of TiO₂ nanoparticles was added in 100ml of distilled water and was thoroughly homogenized in an ultrasonic bath for 30 minutes. 30 ml 0.2mM silver nitrate solution was added in the above. Chilled 30 ml of 0.2 mM sodium borohydride solution was added drop wise in the above solution with continuous stirring on magnetic stirrer. The pH was maintained at 10 adding NH₄OH.

3. Characterization

3.1 UV-Vis Spectroscopy

The Ag doped TiO₂ nanoparticles were characterized by UV- Vis spectroscopy which revealed that the UV Vis exhibiting the typical surface Plasmon absorption maxima at the formation of silver nanoparticles by wavelength in the range of 400-450nm as observe. The highest peak observed is at 435nm in fig 1.

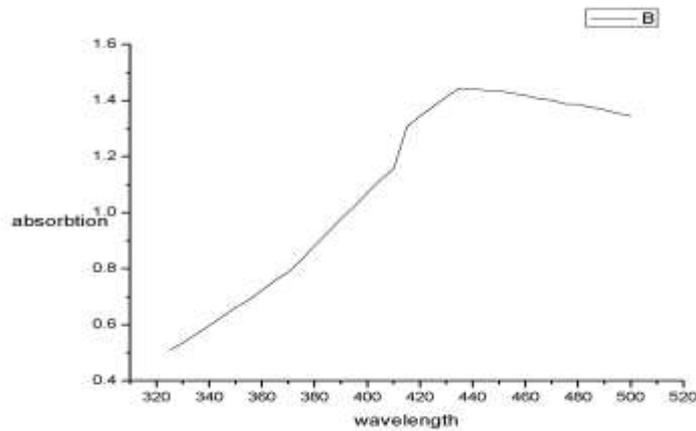


Fig.1. Absorption spectra of silver doped titanium dioxide nanoparticles

3.2 X ray Diffraction

The crystallite size D can be calculated by using Debye-Scherrer formula.

$$D = 0.94 \lambda / \beta \cos \theta$$

where D is the average crystallite domain size perpendicular to the reflecting planes, λ is the X-ray wavelength, β is the full width at half maximum (FWHM), and θ is the diffraction angle. The crystallite size of the synthesized silver doped titanium dioxide nanoparticles obtained by Debye-Scherrer formula is 22 nm and the peakes were matched with the JCPDS file no. 88-1175 as shown in fig. 2.

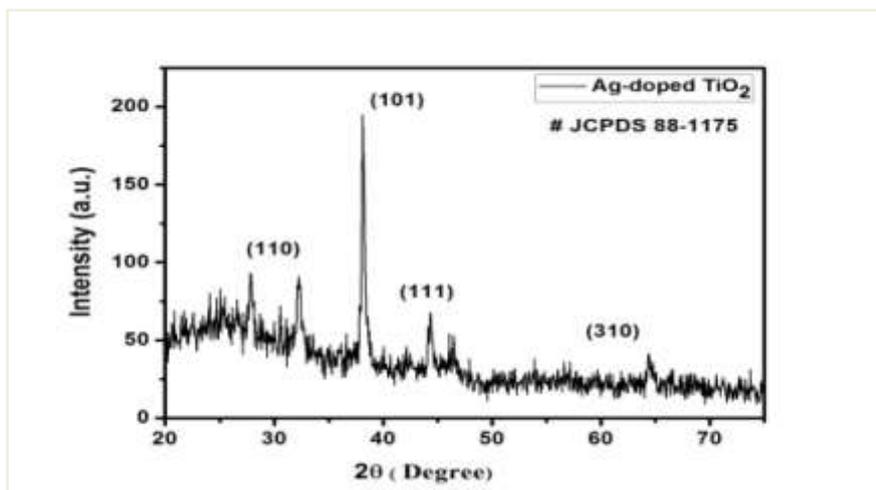


Fig. 2: X-ray diffraction pattern of silver doped titanium dioxide nanoparticles

3.3 Fourier transforms infrared spectroscopy

The common used region for infrared absorption spectroscopy is $4000 \sim 400 \text{ cm}^{-1}$ because the absorption radiation of most organic compounds and inorganic ions is within this region. FTIR spectra of silver doped titanium dioxide nanoparticles have absorption band at 1550 cm^{-1} corresponding to C=C and C-H group, band at 1678 cm^{-1} , 2262 cm^{-1} , 2402 cm^{-1} , 3702 cm^{-1} shows stretching of C-H, C=C, C=N, O-H, N-H groups respectively as shown in fig 3.

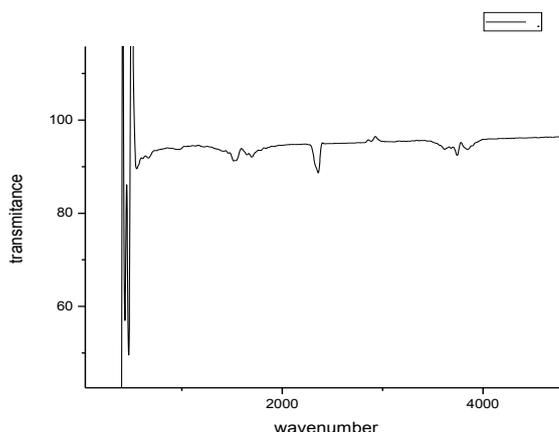


Fig.3. I.R. spectra of silver doped titanium dioxide nanoparticle

3.4 SEM Imaging

SEM analysis was performed by JEOL-JAPAN scanning machine from Aligarh Muslim University. The image shows silver doped titanium nanoparticles. Surface morphology of the particle reveals that the particles are not of equal size rather it is showing clusters of Particle as in fig.4. Edax image shows the Peaks of silver, titanium and oxygen which confirms the Presence of silver doped Titanium dioxide nanoparticle in fig.5.

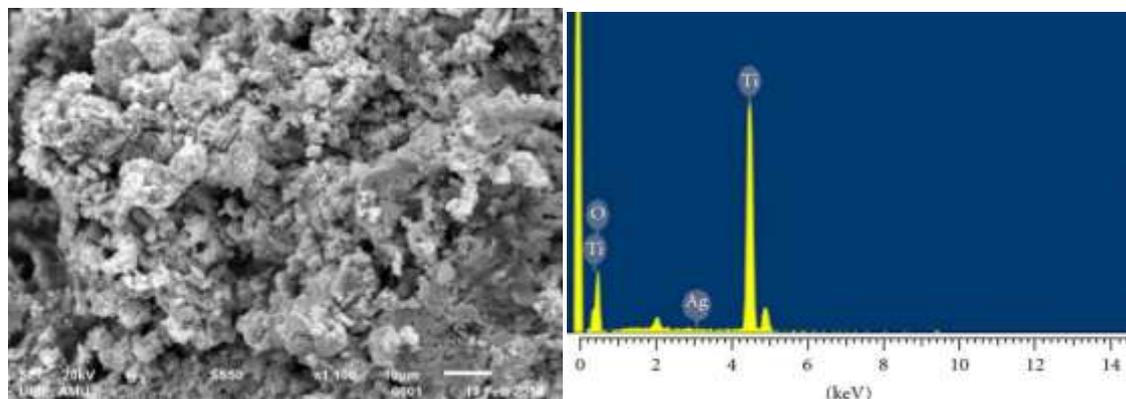


Fig: 4: SEM image of silver doped titanium dioxide nanoparticle

Fig.5: Energy dispersive X-ray analysis of silver doped nanoparticle

4. ANTIBACTERIAL ACTIVITY OF AG/TIO₂.

4.1 Disc diffusion method

The antibacterial activity of the synthesized silver doped titanium dioxide Nanoparticles was tested using the standard disc Diffusion method, leading to the inhibition of bacterial growth. We have used lower concentrations (10, 30, 50, 70 and 90 µg/ml) of Ag/TiO₂ Nps to determine zone of inhibition. We have ciprofloxacin an antibiotic as positive control.

5. Results and Discussion

The quantitative evaluation of Silver doped titanium dioxide nanoparticles showed broad spectrum antibacterial property against *Pseudomonas aeruginosa* as compared to that of conventional antibiotic ciprofloxacin that can be clearly observed by the diameter (in mm) of the zone of growth inhibition of bacteria around the diffused discs. As the concentration of Ag-TiO₂ nanoparticles increased the diameter of inhibitory zone get enhanced, as shown in table 1. The antibiotic ciprofloxacin shows resistance against bacterial pathogens to some extent as compared to that of silver doped titanium dioxide nanoparticles.

Here in fig.6 the graphical representation has clearly explained the inhibitory effect of Ag/TiO₂ nanoparticles.

TABLE: 1

Comparative study of antibacterial activity of silver doped titanium dioxide nanoparticles and antibiotic ciprofloxacin		
	Diameter of zone of inhibition (mm)	
Concentration in(µg/ml)	Ciprofloxacin (positive control)	Ag/TiO ₂
10	1	1
30	2	4
50	2.5	5
70	3	6.5
90	5	7

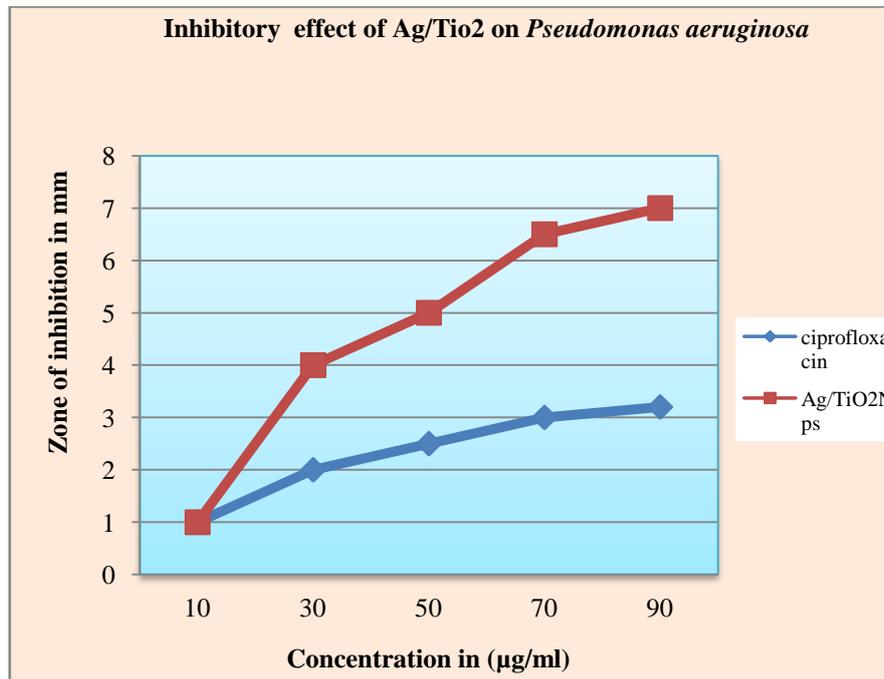


Fig. 6: Graphical represent of growth inhibitory effect of silver doped titanium dioxide nanoparticle

6. CONCLUSION

The importance in medical applications of silver and antibacterial activity of TiO₂ together led researchers to think about the manufacture of silver doped titanium dioxide nanoparticles for food preparation surfaces, air conditioning filters, etc. This paper describes the synthesis of TiO₂ containing different amounts of silver metal by the sol-gel route. Synthesis of silver doped titanium dioxide nanoparticles was confirmed by UV Vis measurement and presence of titanium, silver and oxygen confirmed by peaks of edax. Crystallite size was obtained to be 22 nm. Ag doped TiO₂ nanoparticles show enhanced antibacterial activity rather than as by ciprofloxacin antibiotic. At concentration 90µg/ml the silver doped titanium dioxide nanoparticles show maximum growth inhibition.

7. FUTURE PROSPECTS

Due to multidrug and antibiotic resistance of pathogenic bacteria silver doped titanium dioxide nanoparticles have proved to be good and convenient alternative. In future Ag/TiO₂ nanoparticles can be commercially exploited in pharmaceutical companies, and medicine. Due to antibacterial properties it can be used for water treatment in purifiers. Manufacturing bandages for wound healing.

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